

Preliminary communication

THE SYNTHESIS OF FLUXIONAL RUTHENOCENE BIS-CROWN ETHERS

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Summary

The synthesis of novel ruthenocene bis-crown ethers is reported. Dynamic ^{13}C NMR studies reveal a common intramolecular fluxional phenomenon resulting from rotation about the N—CO bond system.

The design and synthesis of macropolycyclic molecules containing a redox-active centre in close proximity to a coordination site has been the subject of many recent publications [1—4]. Interest in these molecules stems from the idea of studying any interactions between the redox-active centre and a closely bound metal cation guest species.

The metallocene redox centre ferrocene has been successfully appended to a variety of crown ethers and their coordination and redox chemistries are currently under investigation [5, 6]. This communication reports the synthesis of novel ruthenocene bis-crown ethers and of a ruthenocene bis-morpholine amide derivative.

All the compounds were prepared by the standard condensation reaction (Scheme 1) and gave satisfactory elemental analysis, ^1H NMR and molecular masses by mass spectrometry.

The ^{13}C NMR of 4, 5 and 6 were recorded at various temperatures and all three were found to display fluxional behaviour on the NMR timescale. At 223 K the ^{13}C NMR reveals two absorptions for the respective NCH_2 carbons of 4, 5 and 6. On warming these individual signals broaden and eventually coalesce. Also the respective OCH_2 carbon absorptions collapse to give simplified broadened peaks.

These observations suggest that at the coalescence temperatures and above the aza-crown ether rings of 4 and 5, and the morpholine rings of 6 are no longer fixed relative to the respective carbonyl groups and rotation about the amide N—CO bond system is fast on the NMR timescale. This dynamic phen-

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