Journal of Organometallic Chemistry, 306 (1986) C10-C12 Elsevier Sequoia S.A., Lausanne – Printed in The Netherlands

Preliminary communication

THE SYNTHESIS OF FLUXIONAL RUTHENOCENE BIS-CROWN ETHERS

PAUL D. BEER* and A.D. KEEFE

Department of Chemistry, The University of Birmingham, Birmingham B15 2TT (Great Britain) (Received February 10th, 1986)

Summary

The synthesis of novel ruthenocene bis-crown ethers is reported. Dynamic ¹³C NMR studies reveal a common intramolecular fluxional phenomenon resulting from rotation about the N—CO bond system.

The design and synthesis of macropolycyclic molecules containing a redoxactive centre in close proximity to a coordination site has been the subject of many recent publications [1-4]. Interest in these molecules stems from the idea of studying any interactions between the redox-active centre and a closely bound metal cation guest species.

The metallocene redox centre ferrocene has been successfully appended to a variety of crown ethers and their coordination and redox chemistries are currently under investigation [5, 6]. This communication reports the synthesis of novel ruthenocene bis-crown ethers and of a ruthenocene bis-morpholine amide derivative.

All the compounds were prepared by the standard condensation reaction (Scheme 1) and gave satisfactory elemental analysis, ¹H NMR and molecular masses by mass spectrometry.

The ¹³C NMR of 4, 5 and 6 were recorded at various temperatures and all three were found to display fluxional behaviour on the NMR timescale. At 223 K the ¹³C NMR reveals two absorptions for the respective NCH₂ carbons of 4, 5 and 6. On warming these individual signals broaden and eventually coalesce. Also the respective OCH₂ carbon absorptions collapse to give simplified broadened peaks.

These observations suggest that at the coalescence temperatures and above the aza-crown ether rings of 4 and 5, and the morpholine rings of 6 are no longer fixed relative to the respective carbonyl groups and rotation about the amide N—CO bond system if fast on the NMR timescale. This dynamic phen-

0022-328X/86/\$03.50 © 1986 Elsevier Sequoia S.A.





SCHEME 1

TABLE 1

COALESCENCE TEMPERATURES (T_c), CHEMICAL SHIFT DIFFERENCES ($\Delta \nu$) AND $\Delta G^{\pm a}$ VALUES OF 4, 5 AND 6

Compound	T _c (±5 K)	$\Delta \nu$ at 223 K (Hz)	ΔG^{\ddagger}
			$(kJ mol^{-1})$
4	320	68.1	64.1
5	323	238.0	61.4
6	298	1178	54.5

 ${}^{a}\Delta G^{\dagger}$ values calculated from the Gutowsky [7] and Eyring [8] equations.

omenon is also displayed by related ferrocene bis-crown ether molecules [6]. The respective coalescence temperatures (T_c) , chemical shift differences $(\Delta \nu)$

and ΔG^{\dagger} values for this dynamic process are reported in Table 1. It is evident that the ΔG^{\dagger} values for 4 and 5 are very similar and are found to be in good agreement with ΔG^{\dagger} values obtained for the analogous ferrocene bis-crown ethers [6].

The coordination chemistry of these ruthenocene bis-crown ethers and other metallocene crown and cryptand molecules will be described in the full paper.

We thank the SERC for financial support and for use of the high field NMR Service at the University of Warwick.

References

- 1 S. Akabori, Y. Habata, Y. Sakamoto, M. Sato and S. Ebine, Bull. Chem. Soc. Jpn., 56 (1983) 537.
- 2 B. Czech, A. Ratajczak and K. Nagraba, Monatsh. Chem., 113 (1982) 965.
- 3 R. Wolf Jr. and S.R. Cooper, J. Am. Chem. Soc., 106 (1984) 4646.
- 4 T. Izumi, T. Tezuka, S. Yusa and A. Kasahara, Bull. Chem. Soc. Jpn., 57 (1984) 2435.
- P.D. Beer, J. Chem. Soc. Chem. Commun., (1985) 1115.
 P.D. Beer, J. Organomet. Chem., 297 (1985) 313.
- 7 H.S. Gutowsky and C.H. Holm, J. Chem. Phys., 25 (1956) 1228.
- 8 S. Gladstone, K.J. Laidler and H. Eyring, The Theory of Rate Process, McGraw-Hill Book Co. Inc., New York, 1941, Chap. 1.